Pandit Deendayal Petroleum University

School of Technology

BSC 501P					Inorganic Chemistry Lab – I					
Teaching Scheme					Examination Scheme					
L	т	Р	С	Hrs/Week	Theory			Practical		Total
					MS	ES	IA	LW	LE/Viva	Marks
0	0	2	1	2				50	50	100

### **COURSE OBJECTIVES**

- > To apply the theoretical knowledge of Inorganic chemistry into practical application.
- > To demonstrate the skill for quantitative estimation of inorganic compounds.
- > To develop the skills for synthesis of pure inorganic complexes and their crystals.
- > To learn good and safe laboratory practices.

### LIST OF EXPERIMENTS

- 1. Quantitative estimation of Ni<sup>2+</sup> as Ni-dimethyl glyoxime.
- 2. Preparation of cuprous chloride.
- 3. To prepare pure crystals of Tetra amine copper (II) sulphate.
- 4. Determination of amount of Ferrous iron in Mohr's salt by titration against standard KMNO<sub>4</sub> solution.
- 5. Estimation of copper in a given solution.
- 6. Preparation of Prussian blue from iron fillings.
- 7. To prepare pure crystals of chrome alum.
- 8. Estimation of Barium in a salt solution.

9. To determine the percentage purity of the given sample of MgSO<sub>4</sub>.7H<sub>2</sub>O and also determine the percentage of magnesium in it by provided N/20 EDTA solution.

## **COURSE OUTCOMES**

On completion of the course, student will be able to

CO1– Apply the theoretical knowledge of Inorganic Chemistry in quantification & synthesis of inorganic compounds.

CO2– Demonstrate the skills for quantitative estimation of ions viz. Ni<sup>2+</sup>, Cu<sup>2+</sup>, Ba<sup>2+</sup>, Fe<sup>2+</sup> in salt or solution.

CO3– Prepare metal salts and complexes.

- CO4– Learn to prepare pure crystals of inorganic compounds.
- CO5– Determine the percentage purity and percentage of Mg in a given unknown sample.
- CO6– Develop the aptitude for research & development in analytical and synthetic Inorganic Chemistry.

### **TEXT/REFERENCE BOOKS**

- 1. Mendham, J., Vogel's Quantitative Chemical Analysis Sixth Edition, Pearson, 2009.
- 2. Svehala, G. and Sivasankar, B., Vogel's Qualitative Inorganic Analysis, Pearson, India, 2012.
- 3. Marr, G. and Rockett, B. W., Practical Inorganic Chemistry. John Wiley & Sons 1972.

# SEMESTER EXAMINATION PATTERN

#### Max. Marks: 100

LW(Daily lab performance plus journal) LE (Viva-voce plus Lab examination) Exam Duration: 3 Hrs 50 Marks 50 Marks